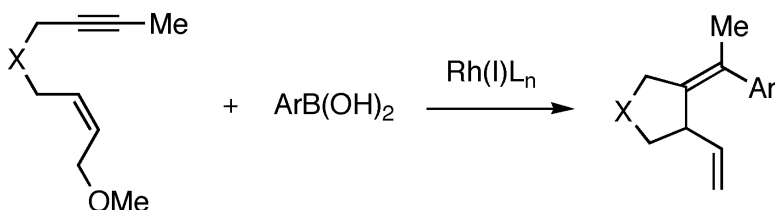


Rhodium-Catalyzed Cyclization of 1,6-Enynes Triggered by Addition of Arylboronic Acids

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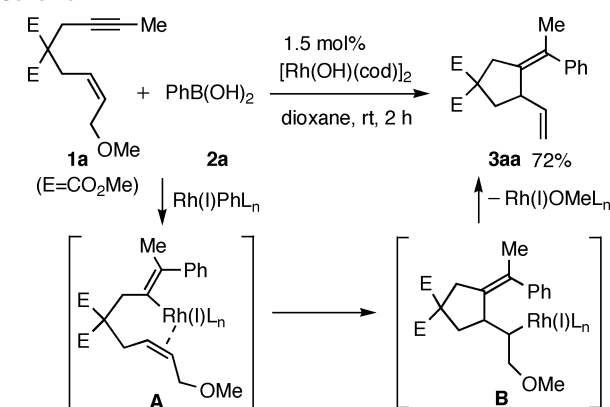
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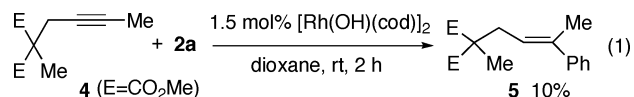
The rhodium(I)-catalyzed addition of arylboronic acids to alkenes or alkynes has recently emerged as a useful synthetic protocol for the formation of carbon–carbon bonds in organic chemistry.¹ Unlike most palladium-catalyzed carbon–carbon bond-forming reactions, which involve a Pd(II)/Pd(0) redox process,² the formal oxidation state of rhodium remains 1+ through the reaction. In the course of the addition reaction, the intermediate organorhodium(I) species is easily protodemetalated by a proton source (HX) that is present as a cosolvent or additive, regenerating the catalytically active Rh(I)–X species to promote the next catalytic cycle.³ The intermediate complexes are rarely used for further carbon–carbon bond formation despite their potential usefulness. Recently a few reports on cyclization reactions involving the second carbon–carbon bond-forming process have appeared, wherein a catalytic rhodium(I) species was also regenerated by protodemetalation.⁴ We then envisaged that the ensuing carbon–carbon bond formation would be feasible in a catalytic sense if an allylic ether were placed at an appropriate position in the molecule. The intermediate organorhodium complex formed by the intramolecular addition to the allylic carbon–carbon double bond may undergo a facile β -alkoxy elimination.⁵ The resulting alkoxyrhodium(I) would be suitable to participate in the next catalytic cycle through transmetalation with an arylboronic acid.⁶ In this report, we describe the intramolecular cyclization reaction of 1,6-enynes that is triggered by the rhodium-catalyzed addition of arylboronic acids to the carbon–carbon triple bond.^{7,8}

When 1,6-enyne **1a** was treated with phenylboronic acid (**2a**, 2.0 equiv) in the presence of [Rh(OH)(cod)]₂ (0.03 equiv of Rh, cod = cycloocta-1,5-diene)^{3a} in dioxane at room temperature under a nitrogen atmosphere for 2 h (*COD conditions*), (*Z*)-1-(1-phenylethylidene)-2-vinylcyclopentane **3aa** was obtained as a single isomer in 72% isolated yield (Scheme 1). The *Z* configuration of the exo double bond was confirmed by a difference NOE study. The reaction is initiated by regioselective addition of a phenylrhodium(I) species, generated in situ by the transmetalation of rhodium(I) with phenylboronic acid (**2a**), onto the alkyne, giving the alkenylrhodium(I) intermediate **A**. Intramolecular carborrhodation to the pendent allylic double bond then occurs in a 5-exo mode, leading to the formation of the alkyrhodium(I) intermediate **B**. Finally, β -elimination of methoxy group affords the (*Z*)-1-(1-phenylethylidene)-2-vinylcyclopentane **3aa** with generation of a catalytically active rhodium(I) methoxide. It is noteworthy that the phenylrhodium species formed by transmetalation underwent highly regioselective 1,2-addition across the carbon–carbon triple bond flanked by methyl and primary alkyl substituents.⁹ When the dppb was used as an additional ligand [0.03 equiv of [Rh(OH)(cod)]₂/dppb, 2 equiv of PhB(OH)₂, dioxane, room temperature (*DPPB conditions*)], 1,6-enyne **1a** gave **3aa** in moderate yield (49%) in 2 h with 32% of **1a** unreacted. For comparison, an analogous reaction was carried out using substrate **4** lacking an olefinic moiety. The addition reaction proceeded sluggishly and in 2 h formed the 1,2-

Scheme 1



adduct **5** only in 10% yield under the *COD* conditions (eq 1). Almost no reaction occurred under the *DPPB* conditions (vide supra). These contrasting results indicated that the olefinic moiety of **1a** intramolecularly coordinates to rhodium to facilitate the initial 1,2-addition.¹⁰ It is also conceivable that the stronger π -acidic character of the cod ligand facilitates intramolecular olefin coordination with **A**.



The results obtained with a variety of arylboronic acids and 1,6-enynes are summarized in Table 1. The catalytic process worked well with a sterically and electronically diverse array of arylboronic acids to give the corresponding products **3ab–3ah** in good yield. Cyclization successfully occurred with substrates having a free hydroxyl group or a silyl ether at the allylic position. The reaction of the allylic acetate was slower and required a larger amount of the rhodium catalyst (6 mol %). A slightly better yield was obtained with **1e**, having an ethyl-substituted alkyne, than with **1a**. The reaction of phenyl-substituted alkyne **1f** suffered from a lower regioselectivity of the initial 1,2-addition to give the product **3fa** only in 25% yield. Enyne **1g**, having an *E*-olefin, was also converted to the product **3aa** in good yield. Even substrate **1h**, equipped with trisubstituted olefin, reacted to give the product **3ha**, bearing a quaternary carbon center. Substrate **1i**, having a dimethyl acetal at the allylic position, gave the aldehyde **3ia** in 70% yield after acidic hydrolysis. The reaction tolerated the presence of a variety of functional groups, including ether **1j** and sulfone **1k**. The reaction of aza-1,6-enyne **1l**, bearing a sulfonamide group in the backbone, gave the product **3la** in only 27% yield due to the low regioselectivity of 1,2-addition of a phenylrhodium(I) species to alkyne.

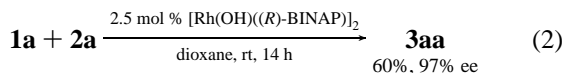
Next, we examined the asymmetric version of this process using a chiral phosphine ligand instead of the cod ligand.^{3a,11} A high level of asymmetric induction (97% ee) was observed with the product

Table 1. Rhodium-Catalyzed Cyclization of 1,6-Enynes **1** with Arylboronic Acids **2**^a

| substrate ^b | ArB(OH) ₂ | product | time (h) | yield (%) ^c |
|---|--|------------|----------|------------------------|
| | | | | |
| 1a R=Me R'=Me | 2b 4-F-C ₆ H ₄ | 3ab | 2 | 77 |
| 1a | 2c 4-NO ₂ -C ₆ H ₄ | 3ac | 6 | 82 |
| 1a | 2d 4-Me-C ₆ H ₄ | 3ad | 10 | 68 |
| 1a | 2e 3-MeO-C ₆ H ₄ | 3ae | 2 | 77 |
| 1a | 2f 3-Cl-C ₆ H ₄ | 3af | 2 | 77 |
| 1a | 2g 2-Me-C ₆ H ₄ | 3ag | 8 | 80 ^d |
| 1a | 2h 1-naphthyl | 3ah | 13 | 81 ^d |
| 1b R=Me R'=H | 2a Ph | 3aa | 2 | 63 |
| 1c R=Me R'=TBS | 2a Ph | 3aa | 2 | 78 |
| 1d R=Me R'=Ac | 2a Ph | 3aa | 24 | 52 ^{e,f} |
| 1e R=Et R'=Me | 2a Ph | 3ea | 2 | 80 |
| 1f R=Ph R'=Me | 2a Ph | 3fa | 2 | 25 ^g |
| | | | | |
| 1g R=H | 2a Ph | 3aa | 2 | 75 |
| 1h^h R=Me | 2a Ph | 3ha | 23 | 61 |
| | | | | |
| 1i | 2a Ph | 3ia | 2 | 70 |
| | | | | |
| 1j X=C(CH ₂ OMe) ₂ | 2a Ph | 3ja | 2 | 81 |
| 1k X=C(SO ₂ Ph) ₂ | 2a Ph | 3ka | 16 | 70 ^e |
| 1l X=NTs | 2a Ph | 3la | 12 | 27 ^e |

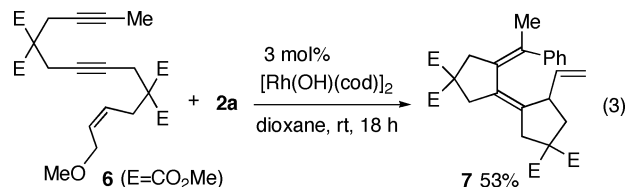
^a The reaction was carried out with **1** (0.2 mmol) and **2** (0.4 mmol) in dioxane (2 mL) in the presence of [Rh(OH)(cod)]₂ (0.03 equiv of Rh) at room temperature, unless otherwise noted. ^b E = CO₂Me. ^c Isolated yields. ^d Mixture of atropisomers (52:48 for **3ag**, 62:38 for **3ah**). ^e **2** (4 equiv) and [Rh(OH)(cod)]₂ (0.06 equiv of Rh) were used. ^f The starting material remained. ^g 50 °C. ^h (E):(Z) = 9:1.

3aa, whereas the chemical yield was moderate owing to a low regioselectivity that was peculiar to the use of phosphine ligands (eq 2).¹² We assume that the less sterically demanding character of the cod ligand compared with (*R*)-BINAP contributes to the high regioselectivity observed under the COD conditions.



Thus, the intermediate organorhodium(I) species was successively utilized in the second carbon–carbon bond formation as part of the catalytic cycle. Finally, we examined a domino cyclization process using enediyne **6**.¹³ When **6** was treated with phenylboronic

acid (**2a**) in the presence of [Rh(OH)(cod)]₂ (0.06 equiv of Rh) for 18 h, the bicyclic triene derivative **7** was obtained in 53% yield as a single isomer through successive carboration processes (eq 3).



In summary, we have developed new cyclization reactions of 1,6-enynes triggered by the rhodium-catalyzed regioselective addition of an arylrhodium(I) species to alkynes. The results obtained demonstrated that multiple carbon–carbon bond-forming processes can operate with a single catalytic system using only rhodium(I).

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Supporting Information Available: Experimental details and selected spectral data for new compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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